STRUCTURE OF PONASTEROSIDE A, A NOVEL GLYCOSIDE OF INSECT-MOULTING SUBSTANCE FROM PTERIDIUM AQUILINUM VAR. LATIUSCULUM

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We now wish to report the occurrence of a glycoside of an insect-moulting steroid in Nature. This substance was isolated from <u>Pteridium aquilinum</u> Kuhn var. <u>latiusculum</u> Underw. (Pteridiaceae), found to show moulting hormone activity, and named warabisterone.¹⁾ However, since further study has revealed it to be a glycoside, we have changed the term to ponasteroside A after its aglycone ponasterone A (vide infra).

Ponasteroside A, m.p. 278~279.5°, $[\alpha]_{D}$ +28.5° (pyridine), shows positive color reactions for steroid. The molecular formula was established to be $C_{33}H_{54}O_{11}$ by elemental analysis and the appearance of the molecular ion peak at m/e 626 in the mass spectrum. Its spectral properties are very similar to those of the common insect-moulting substances, e.g., ponasterone A (II) $\frac{2-4}{2}$ Thus the presence of many hydroxyl groups (ν_{max} 3430 cm⁻¹) and a 14-hydroxy-7-en-6-one system in the 5β-steroid nucleus (λ_{max} 245 mµ shifted to 298 and 244 mµ after acid treatment, ν_{max} 1650 cm⁻¹, 1H doublet at 6.18 p.p.m^{*}, $[\theta]_{339}^{max}$ 45 × 10³ (dioxan)) is indicated. The NMR spectrum shows five methyl signals whose chemical shifts and splitting patterns are compatible with those of ponasterone A (II) with the exception that the C-19 methyl proton signal of ponasteroside A is shifted upfield by 0.15 p.p.m. than that of ponasterone A (Table I). At this point it was expected that ponasteroside A is closely related to ponasterone A. However, the composition indicates the presence of $C_{6}H_{10}O_{5}$ more than in ponasterone A, and this increment may be corresponding to the linkage of a common hexose. In consistent with this view, the NMR spectrum exhibits that ponasteroside A possesses carbinyl hydrogens much more than ponasterone A. Based on the above observations it was quite reasonable to consider that ponasteroside A might be a glycoside of ponasterone A.

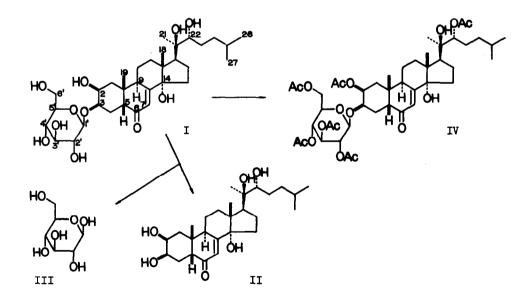
Enzymatic hydrolysis of ponasteroside A was then carried out to give ponasterone A (II), m.p.

TABLE I. Methyl chemical shifts (pyridine).

		C-18	C-19	C-21	C-26	C-27
Ponasterone A	(II) ²⁾	1.16	1.03	1.51	0.82d	0.82d
Ponasteroside A	(I)	1.17	0.88	1.54	0.84d	0.84d

Proton signals (CDC13, 100 MHz). TABLE II. C-2α C-3α C-7 C-9 C-18 C-19 C-21 C-22 C-26 C-27 Ponasterone A 2,3,22-tri-acetate⁵⁾ 5.05 5.32 5.86 3.12 0.85 1.02 4.82 0.88 0.88 1.24 ddd ddd d ddd dd d d s 8 8 Ponasteroside A 2,22,2',-~4.8 -4.15 5.82 3.05 0.85 0.96 1.24 -4.8 0.88 0.88 3', 4', 6'-hexaacetate ¥ ¥ đ ddd 8 s 8 d d C-1' C-2' C-31 C-4' C-5 ' C-6' 4.59 d 5.22 5.04 4.10 Stigmastanyl β -D-glucoside 4.93 3.64 4.24 dd ddd dd dd dd dd 2', 3', 4', 6'-tetraacetate -5.02 5.22 Ponasteroside A 2,22,2',-4.46 ~5.02 3.65 4.05 4.24 d dd ada ad d d 3',4',6'-hexaacetate

* Patterns are unclear due to overlapping of the signals.



277~278°, and glucose (III), establishing that ponasteroside A is a glucoside of ponasterone A.

As described above, the fact that in the NMR spectrum of ponasteroside A the chemical shift of the C-19 methyl signal is displaced as compared with that of ponasterone A suggests glucose to be linked with ponasterone A at ring A. In order to establish the environment of the glucoside linkage, ponasteroside A was acetylated to give the hexaacetate (IV) whose NMR spectrum was compared with the spectra of ponasterone A triacetate⁵⁾ and stigmastanyl B-D-glucoside tetraacetate (Table II). Two signals originating from the two carbinyl hydrogens in ring A of the hexaacetate (IV) appear at ~4.15 and ~4.8 p.p.m. From the line positions, the former signal must be ascribable to a hydrogen on a carbon attached to an oxygen participating the glucoside linkage. and the latter signal attributable to a hydrogen on an acetoxyl-bearing carbon. Although overlapping of the signals does not permit the exact determination of the band width at half height of each signal, the signal shape of the former (~4.15 p.p.m.) indicates that it is associated with an equatorial-like hydrogen, while that of the latter (~4.8 p.p.m.) shows that it is due to an axial-like hydrogen. On the other hand, a chair conformation of ring A of the acetate (IV) was derived from the findings that the ORD curves of ponasteroside A and its acetate (IV) show a similarity to those of ponasterone A (II) and its triacetate, and that an intramolecular nuclear Overhauser effect was observed between the carbinyl hydrogen (~4.8 p.p.m.) and the C-9 allylic hydrogen (3.05 p.p.m.) in the acetate (IV). The combined evidence leads to the conclusion that in ponasteroside A the aglycone is bonded through the C-3 oxygen to glucose.

The C-1' anomeric hydrogen signals in the NMR spectra of ponasteroside A and its acetate (IV) appear as doublets with the large coupling constants (both 7 Hz), demonstrating that glucose is combined in the β -form with the aglycone.

On the basis of the above results, ponasteroside A is concluded to be ponasterone A 3- β -glucoside (I).

Of particular interest biologically is the fact that ponasteroside A, a glycoside, also exhibits high insect-moulting hormone activity in the <u>Sarcophaga</u> test and high stimulating effect on protein synthesis in mouse¹⁾ though the possibility that ponasteroside A reveals the activities only after enzymatic hydrolysis in animals, cannot be excluded.

We have hitherto been convinced that insect-moulting substances which occur in the plant kingdom must exist partly as glycosides. Ponasteroside A presently characterized constitutes the first example.

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FOOTNOTE AND REFERENCES

- NMR spectra of the ecdysterols and the acetates were recorded on a Varian HA-100 spectrometer in C₅D₅N and CDCl₃ solution, respectively. Chemical shifts are given in p.p.m. downward from TMS as internal reference. Abbreviations: s=singlet, d=doublet, and dd=doublet of doublets.
- 1) T. Takemoto, S. Arihara, Y. Hikino, and H. Hikino, Chem. Pharm. Bull. (Tokyo), 16, 762 (1968).
- 2) K. Nakanishi, M. Koreeda, S. Sasaki, M. L. Chang, and H. Y. Hsu, Chem. Comm., 1966, 915.
- 3) H. Moriyama and K. Nakanishi, Tetrahedron Letters, 1968, 1111.
- 4) G. Hüppi and J. B. Siddall, Tetrahedron Letters, 1968, 1113.
- 5) K. Nakanishi, private communication.